

## Chapter 1, THE ORGANIZATION OF MATTER

-Matter is anything that has \_\_\_\_\_ and \_\_\_\_\_

- \_\_\_\_\_ is based on the idea that matter is made up of small particles.

-Phases of Matter:

- Solid \_\_\_\_\_ have a \_\_\_\_\_ force of attraction. They have a definite \_\_\_\_\_ and \_\_\_\_\_
- Liquid \_\_\_\_\_ have a \_\_\_\_\_ force of attraction. They have a definite \_\_\_\_\_ but no definite \_\_\_\_\_.
- Gaseous \_\_\_\_\_ are not bound by a \_\_\_\_\_. They don't have any definite \_\_\_\_\_ or \_\_\_\_\_.

-Mixtures contain 2 or more different \_\_\_\_\_ or \_\_\_\_\_.

- Heterogeneous VS Homogeneous (colloids or solutions)  
\_\_\_\_\_ ( \_\_\_\_\_ )
- Solutions (Solute vs. Solvent)
  - Concentration, C (units: \_\_\_\_\_ or \_\_\_\_\_ or \_\_\_\_\_ or \_\_\_\_\_) (FORMULA  $C = \frac{m}{V}$ )
  - Dilution (\_\_\_\_\_ C by adding solvent and \_\_\_\_\_ total Volume of solution) (FORMULA \_\_\_\_\_)
  - Solubility: Saturation is the maximum \_\_\_\_\_ of \_\_\_\_\_ in solution. If more solute is added, a \_\_\_\_\_ will be seen. For solids (like salt), increasing the temperature will \_\_\_\_\_ the solubility (\_\_\_\_\_ saturation conc.). For gases, increasing the temperature will \_\_\_\_\_ the solubility (\_\_\_\_\_ saturation conc.)
  - \_\_\_\_\_ is a substance that is soluble in water, while \_\_\_\_\_ is a substance that is soluble in oil.
- Separating Mixtures
  - Decant: \_\_\_\_\_
  - Centrifuge \_\_\_\_\_
  - Filtration \_\_\_\_\_
  - Evaporation \_\_\_\_\_
  - Distillation \_\_\_\_\_
  - Chromatography \_\_\_\_\_

-Pure Substances contain only one \_\_\_\_\_.

- Elements can be ATOMS, such as \_\_\_\_\_ or MOLECULES such as \_\_\_\_\_

- Characteristic Physical Properties:

- \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_

- Characteristic Chemical Properties

- Litmus paper:
  - Cobalt chloride paper
  - Limewater
  - glowing splint,
  - burning splint
  - open flame

## Chapter 2, THE POWER OF ENERGY

-Energy is the capacity to do \_\_\_\_\_ or produce \_\_\_\_\_. The unit of measurement of energy we use is the \_\_\_\_\_.

-Forms of Energy...

- Give an example for each:

Solar	Elastic	Electrical
Thermal	Radiant	Chemical
Mechanical	Wind	Sound
Hydraulic	Nuclear	

- **Thermal Energy** is energy that comes from the \_\_\_\_\_ motion of the \_\_\_\_\_ that make up a substance.
- **Radiant Energy (light)** is contained and transported in \_\_\_\_\_ waves (shorter wavelength → \_\_\_\_\_ energy)
- **Chemical Energy** is stored in the \_\_\_\_\_ of a molecule.
- **Mechanical Energy** is the \_\_\_\_\_ and mass of an object in its surroundings.

-Energy Transformation means a change of FORM.

- Example, a baseball pitcher: The \_\_\_\_\_ energy in the muscles of the pitcher's body is **transformed** into \_\_\_\_\_ energy in the movement of his/her body.

-Energy Transfer is the same but a different PLACE or OBJECT

- Example 1, the same baseball pitcher: The mechanical energy in the movement of his/her body is **transferred** to the \_\_\_\_\_ that is thrown.
- Example 2: The thermal energy (heat) from a stove is **transferred** to the pot of water heating up to a boil.

-Physical Changes (phase change)

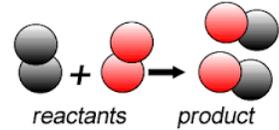
_____ is solid to liquid and energy is _____.	_____ is solid to gas and energy is _____.	_____ is liquid to gas and energy is _____.
_____ is gas to liquid and energy is _____.	_____ is gas to solid and energy is _____.	_____ is liquid to gas and energy is _____.

- **Dissolution** is the creation of a \_\_\_\_\_ by a \_\_\_\_\_ dissolving in a \_\_\_\_\_.
- Some solutes need energy to be dissolved into a solvent, other solutes release energy when dissolved. When dissolution releases E, temperature \_\_\_\_\_. When dissolution absorbs E, temperature \_\_\_\_\_.

- **Deformation** means changing the \_\_\_\_\_ of a material.

-**Chemical Changes** changes the \_\_\_\_\_ and \_\_\_\_\_ of matter.

- **Synthesis** reactions (like photosynthesis) make \_\_\_\_\_ molecules from atoms or simpler molecules. These reactions usually \_\_\_\_\_ energy but can also sometimes \_\_\_\_\_ energy.
- **Decomposition** makes \_\_\_\_\_ molecules or atoms from more complex molecules (breaks them apart). These reactions usually release energy.
- **Oxidation** (ex: rusting metal) is a chemical reaction involving oxygen.
- **Precipitation** is the formation of a solid following the mixing of two solutions. (Ex: vinegar mixed with milk causes solid curds to form).



-Chemical Equations

- The total # of atoms from each element need to be the same in the \_\_\_\_\_ as in the \_\_\_\_\_ (Rule of conservation of mass)

Ex:  $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$  see if this obeys the Rule.

-Electromagnetic Spectrum

- \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_, Visible light, \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_  
 \_\_\_\_\_

### **Chapter 3, THE BEHAVIOUR OF FLUIDS**

-A **Fluid** is a substance that has the capacity to \_\_\_\_\_ and assume the \_\_\_\_\_ of the container into which it has been poured.

-A fluid can be a \_\_\_\_\_ or a \_\_\_\_\_.

-A **compressible fluid** is a fluid whose volume \_\_\_\_\_ change. \_\_\_\_\_ are compressible fluids.

-An **incompressible fluid** is a fluid whose volume \_\_\_\_\_ change. \_\_\_\_\_ are incompressible fluids

-**Pressure** is a \_\_\_\_\_ applied \_\_\_\_\_ to a surface. FORMULA:                      UNITS:

-If force increases, pressure \_\_\_\_\_. If force decreases, pressure \_\_\_\_\_.

-If surface area increases, pressure \_\_\_\_\_. If surface area decreases, pressure \_\_\_\_\_.

-The pressure exerted on an object by an incompressible fluid (like water) depends on:

- The \_\_\_\_\_ of the object in the fluid.
- The \_\_\_\_\_ of the fluid.

-For a compressible fluid, if pressure \_\_\_\_\_ volume decreases, and if pressure \_\_\_\_\_ volume increases.

-3 General Principles

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## **Chapter 4, THE PERCEPTION OF LIGHT AND SOUND**

-A **wave** is a \_\_\_\_\_ that travels through a medium. A wave transports \_\_\_\_\_; it does not transport \_\_\_\_\_.

-**Longitudinal Wave**: moves \_\_\_\_\_ to the motion of it's medium. Example: Sound waves, a slinky.

- Divided into regions of \_\_\_\_\_ and regions of \_\_\_\_\_.

-**Sound Waves** are longitudinal waves produced by the \_\_\_\_\_ of an object and transmitted to the object's environment.

- The Decibel Scale measures the relative intensity of sounds to human ears.
- For every 10dB sound increases \_\_\_\_\_, not \_\_\_\_\_
- EX:
- Sounds above \_\_\_\_\_ are damaging to the human ear.
- Frequency of sound waves (UNIT: \_\_\_\_\_) determines if we can hear it or not. Humans can hear 20 - 20 000Hz. Sounds under 20Hz are called \_\_\_\_\_. Sounds over 20 000Hz are called \_\_\_\_\_

-**Transverse wave**: moves \_\_\_\_\_ to the motion of it's medium. Examples: water surface, electromagnetic waves (Radio → Gamma).

- Wavelength: \_\_\_\_\_ \_\_\_\_\_ means more energy
- Amplitude: \_\_\_\_\_
- Frequency: \_\_\_\_\_ \_\_\_\_\_ means more energy

-**Light** is an electromagnetic wave that is visible to the human eye. (p.111-116)

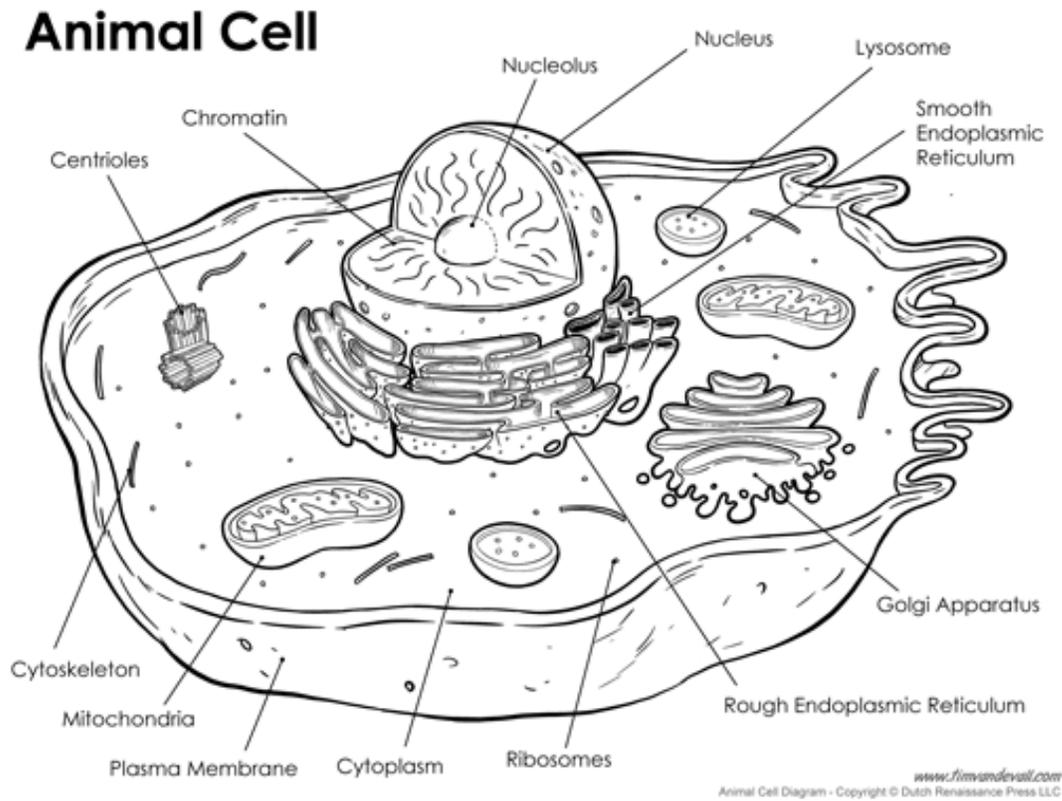
- Lenses refract **light** (cause the wave to deviate or change \_\_\_\_\_)
- There are 2 types of lenses:
  - **Converging**: lenses that cause light to focus on the other side.
    - Ex: \_\_\_\_\_, \_\_\_\_\_
    - Corrects hyperopia (farsightedness) and hyperopia (aging eye muscles)
    - Produces an image that is \_\_\_\_\_, \_\_\_\_\_, and \_\_\_\_\_.
  - **Diverging**: lenses that cause light to separate apart on the other side
    - Corrects myopia (nearsightedness)
    - Produces an image that is \_\_\_\_\_, \_\_\_\_\_, and \_\_\_\_\_.

**Chapter 5, THE PERPETUATION OF LIFE**

**\*\*\*study the diagrams in your textbook\*\*\***

-The CELL is \_\_\_\_\_.

-Cell Structures and Organelles:



Nucleus	
Endoplasmic Reticulum	
Ribosome	
Cytoplasm	

Mitochondria	
Golgi Apparatus	
Lysosome	
Cell Membrane	

-DNA is a chain of \_\_\_\_\_ (of nucleotides).

- DNA bases are A (adenine), C (cytosine), G (guanine), and T (Thymine). They are always paired as \_\_\_\_\_ and \_\_\_\_\_
- The chain is twisted to have a \_\_\_\_\_ structure.
- DNA molecules are extremely long and have to be packed as \_\_\_\_\_ when the cell is dividing.
- Human somatic or body cells have 46 chromosomes (each with a duplicate sister chromatid, connected at the \_\_\_\_\_), that's 23 from the \_\_\_\_\_ parent and 23 from the \_\_\_\_\_ parent. We call these diploid or \_\_\_\_\_.

**-Cellular Specialization** (tissue → organ → system)

- **Tissues:** \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_, \_\_\_\_\_
- **Organs** are made of 2 or more tissue types (ex: the stomach uses all 4 types)
- A **system** is a group of organs and tissues with a common function. (Ex: digestive system from the mouth to the esophagus to the stomach etc...)

**-The Cell Cycle** is separated into \_\_\_\_\_ (most of the cell's life and when normal cell metabolism occurs) and the 4 stages of cell division...

**-Cell Division** (for Mitosis, Meiosis I or Meiosis II)

- \_\_\_\_\_: DNA is packaged as chromosomes and the nuclear membrane disappears.
- \_\_\_\_\_: Chromosomes (Mitosis) or homologous pairs of chromosomes (Mitosis I) align at the center of the cell.
- \_\_\_\_\_: Chromosomes or Homologous pairs are pulled apart by spindle fibres.
- \_\_\_\_\_: New nuclear membranes are formed and cytokinesis occurs (movement of the cytoplasm into 2 new cells)

**-Meiosis** produces \_\_\_\_\_ (n) sex cells. It occurs in the \_\_\_\_\_ (males) and in the \_\_\_\_\_ (females). Two cell divisions produce 4 haploid \_\_\_\_\_ from 1 diploid cell.

- Male gametes are called: \_\_\_\_\_
- Female gametes are called: \_\_\_\_\_

**-Stages of Human Development**

Before Birth \_\_\_\_\_ → \_\_\_\_\_ → \_\_\_\_\_ →

After Birth \_\_\_\_\_ → early childhood → \_\_\_\_\_ → adolescence → \_\_\_\_\_ → old age

**-Puberty Hormones**

- The pituitary gland of the brain releases FSH and LH
  - Primary Sexual Characteristics (Male): \_\_\_\_\_

- Primary Sexual Characteristics (Female): \_\_\_\_\_
- The Testicles produce testosterone (male) and ovaries produce estrogen and progesterone (female)
  - Secondary Sexual Characteristics (Male): \_\_\_\_\_
  - Secondary Sexual Characteristics (Female): \_\_\_\_\_

-Female Reproductive Cycle

- Ovarian Cycle
  - a. In the ovaries, inside an ovarian follicle, a \_\_\_\_\_ oocyte undergoes meiosis I.  
 \*\*Hormone: Triggered by FSH, then ovaries produce \_\_\_\_\_ which stimulate the pituitary to produce more FSH and LH
  - b. In **“Ovulation”**: the \_\_\_\_\_ ovum is expelled into the fallopian tube.  
 \*\*Hormones: triggered by \_\_\_\_\_ in LH. Accompanied by a surge in \_\_\_\_\_ and \_\_\_\_\_.
  - c. Ovarian Follicle turns into the **corpus luteum**  
 \*\*Hormones: The corpus luteum secretes \_\_\_\_\_ (the “STOP signal which inhibits FSH and LH at the \_\_\_\_\_)
  - d. As the corpus luteum disintegrates and \_\_\_\_\_ decreases the pituitary can be produce \_\_\_\_\_ and \_\_\_\_\_ again.
- Menstrual cycle
  - a. Bleeding occurs do to the expulsion of the \_\_\_\_\_ and the unfertilized ovum.  
 \*\*Hormones: Caused by the \_\_\_\_\_ in progesterone.
  - b. Endometrium Thickens  
 \*\*Hormones: Production of \_\_\_\_\_ by a new ovarian follicle.
  - c. Endometrium continue to Thicken.  
 \*\*Hormones: Secretion of \_\_\_\_\_ by the corpus luteum.

-Male Reproductive Cycle:

- Spermatogenesis occurs in the \_\_\_\_\_ of the \_\_\_\_\_
- \_\_\_\_\_ spermatogonium undergo meiosis to produce \_\_\_\_\_ spermatozoa.
- Sperm then travel to the
  - Epididmis (in the \_\_\_\_\_)
  - Vas Deferens
  - Ampulla of the Vas Deferens
    - Seminal Vesicle
  - Urethra
    - Prostate gland
    - Bulbourethral gland

